

# Shamokin Creek Watershed Mine Drainage Overview GEOL 305

Bucknell University

March 2020

Carl Kirby

The Shamokin Creek watershed has the highest percentage of its waters degraded by AMD in PA. AMD degrades the aquatic ecosystem and impairs uses of the entire length of Shamokin Creek to its mouth 15 mi downstream from the mined part of the watershed. AMD contamination mainly results from excessive iron deposition on the streambeds or high acidity. Among the mine discharges, there is a wide range of chemistry and flow conditions. Iron concentration ranges from 0 to 90 mg/L, Mn from 0 to 7 mg/L, Al from 0 to 15 mg/L, and pH ranges from 2.8 to 7. For more information, see Cravotta, C. A. III, and Kirby, C. S., 2004, *Effects of Abandoned Coal-Mine Drainage on Streamflow and Water Quality in the Shamokin Creek Basin, Northumberland and Columbia Counties, Pennsylvania, 1999-2001*, US Geological Survey Water-Resources Investigations Report 03-4311.



U.S. Department of the Interior  
U.S. Geological Survey

**Effects of Abandoned Coal-Mine Drainage  
on Streamflow and Water Quality in the  
Shamokin Creek Basin,  
Northumberland and Columbia Counties,  
Pennsylvania, 1999-2001**

*by Charles A. Cravotta III and Carl S. Kirby*

Water-Resources Investigations Report 03-4311

# Shamokin Creek Watershed & Mining Region of Watershed

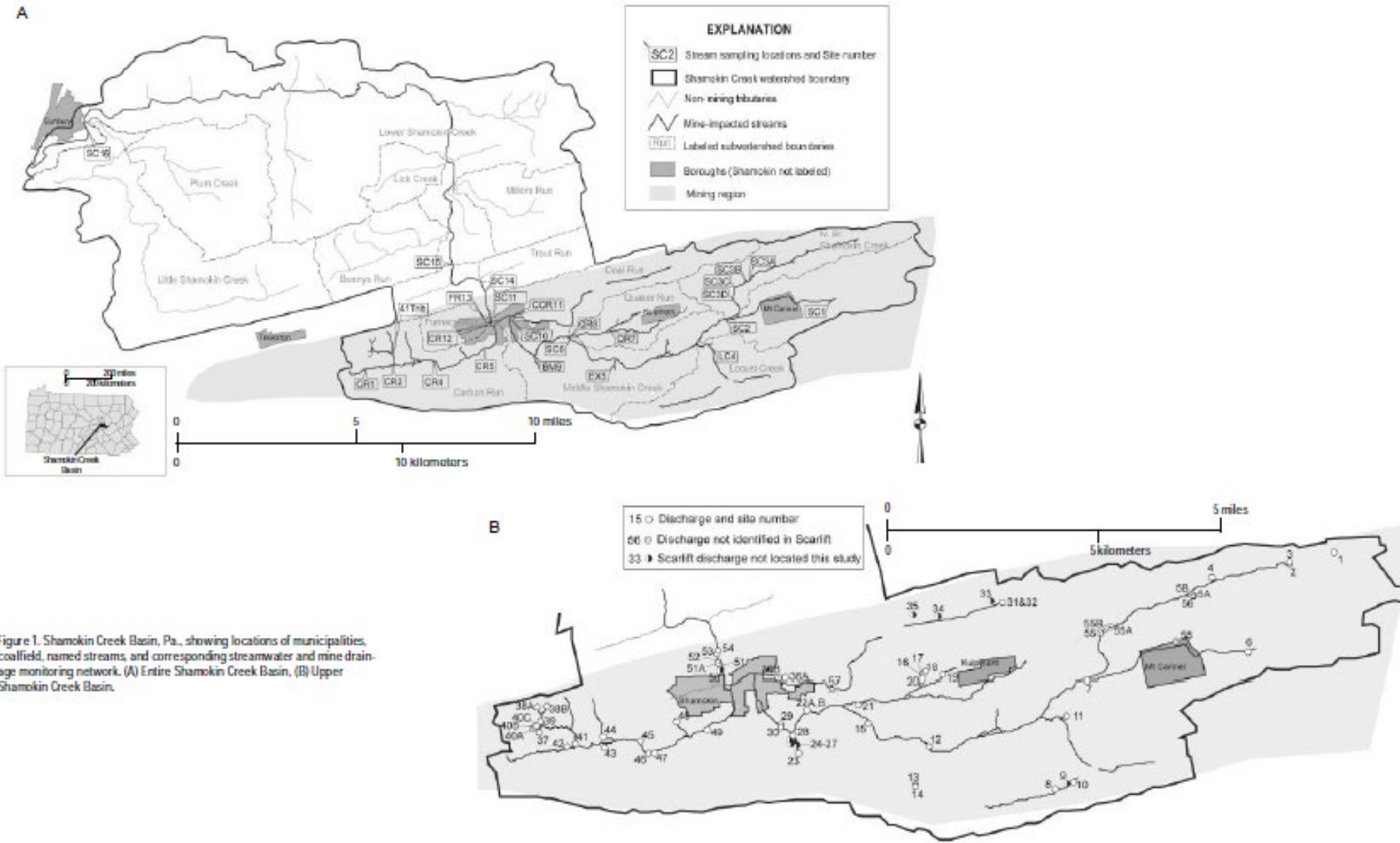


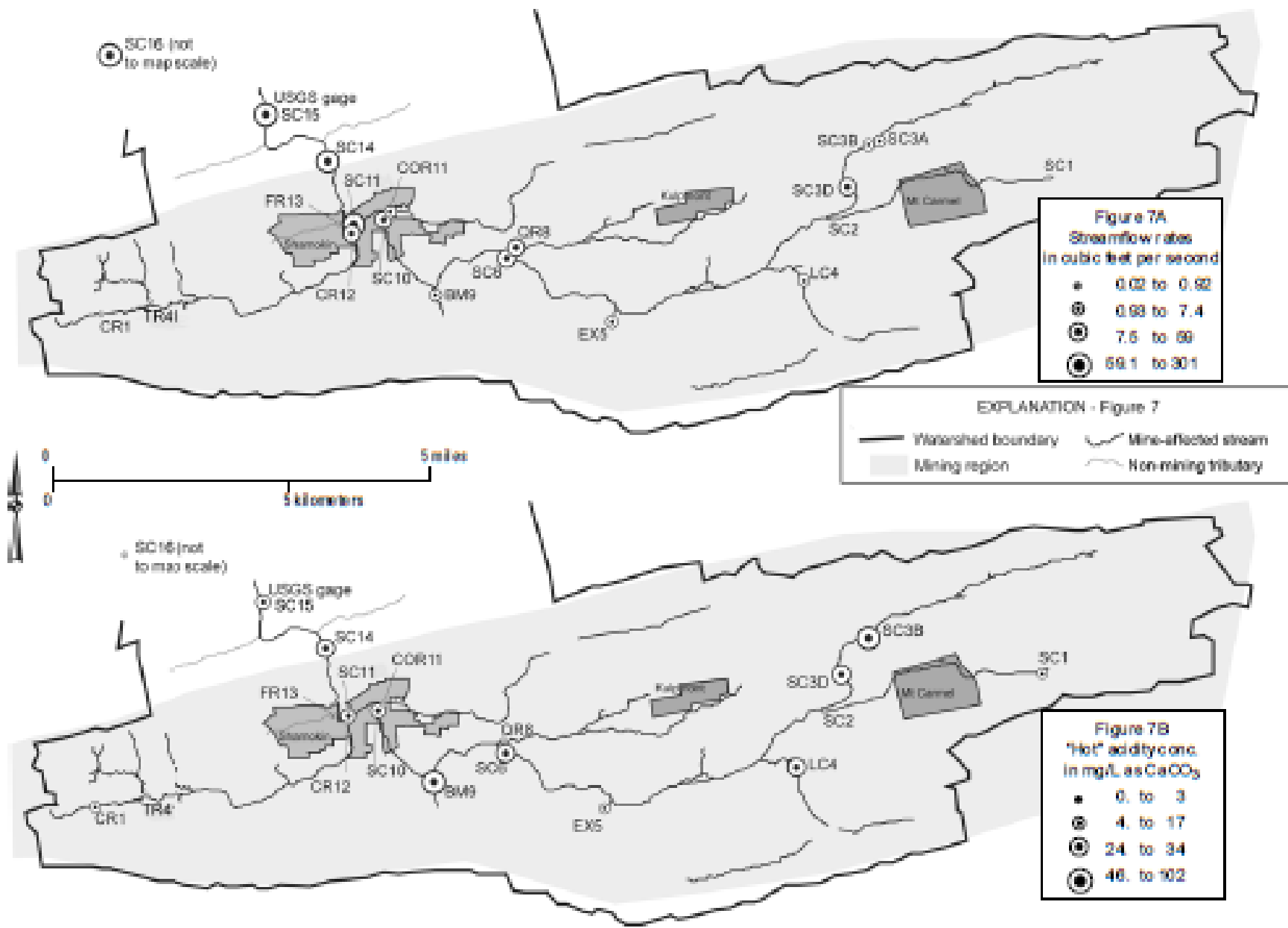
Figure 1. Shamokin Creek Basin, Pa., showing locations of municipalities, coalfield, named streams, and corresponding streamwater and mine drainage monitoring network. (A) Entire Shamokin Creek Basin, (B) Upper Shamokin Creek Basin.

12 Stream sites sampled this study

> 50 “legally abandoned” deep mine or spoils pile discharges of AMD; no responsible party to clean up

USGS Study:

- pH
- Redox
- DO
- flowrates
- Alk, Hot Acidity (= net acidity)
- Samples for metals
- Calc. metal and acidity loading
- Electrofishing 2 sites
- Compare to historical data
- Suggestions for treatment



4 treatment systems  
1 working well (Site 15)

Figure 7. Maps of Shamokin Creek, Pennsylvania, showing: (A) stream flow in cubic feet per second, (B) acidity, (C) dissolved sulfate, (D) dissolved iron, (E) dissolved manganese, and (F) dissolved aluminum in the mainstem and tributaries, March 2000.

Historical (blue) and this study (black) flowrates at discontinued USGS gaging station

Two of the study flowrates were  $> 1$  std. dev. below the 50-year average, so caution must be used when considering treatment based on mass loading.

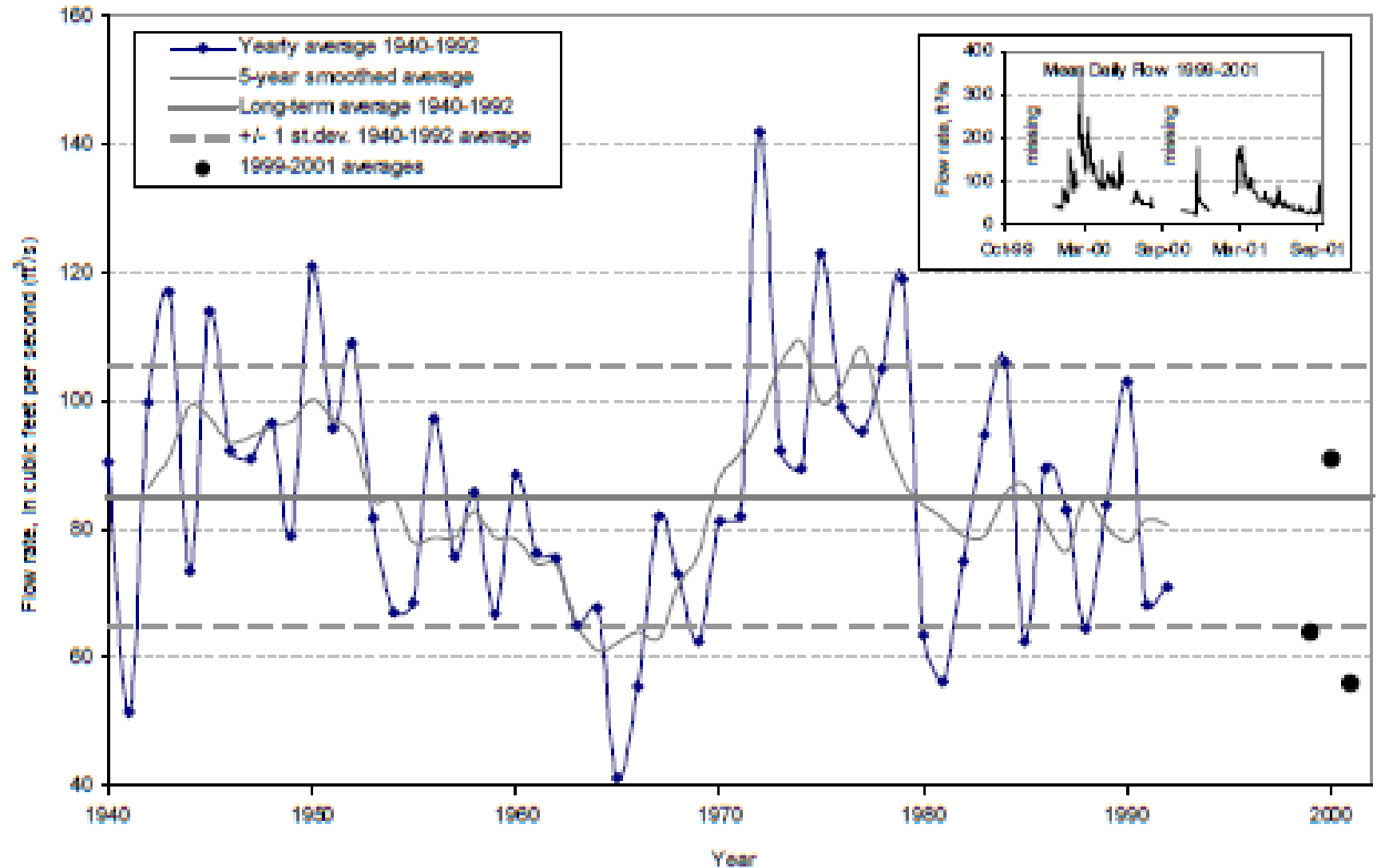


Figure 8. Historical (1940-1992) and recent (1999-2001) streamflow data for Shamokin Creek near Shamokin, Pennsylvania (SC15; USGS station 01554500). Daily mean values were used to generate the hydrographs and to compute the mean and standard deviation (stdev.) of the long-term average streamflow.

# Ranked remediation strategies

**Table 7. Rankings and possible remedial alternatives for abandoned mine drainage (AMD) in Shamokin Creek Basin, Pennsylvania**  
 [AMD rank based on instantaneous dissolved metals, net-alkalinity, or sulfate loading during March 14-16, 2000. Remedial alternatives identified in order of preference; any treatment design would require additional data and specific analysis; VFCW, vertical-flow compost wetland; ALD, anoxic limestone drain; OLD, flowable oxic limestone drain; OLC, open limestone channel; FT/s, cubic feet per second; L/min, liters per minute; mg/L, milligrams per liter; -, no data; =, equal to; <, less than; ≤, less than or equal to; >, greater than; ≥, greater than or equal to. \*Discharges ranked 43rd are not distinguished from one another by rank because of insufficient data.

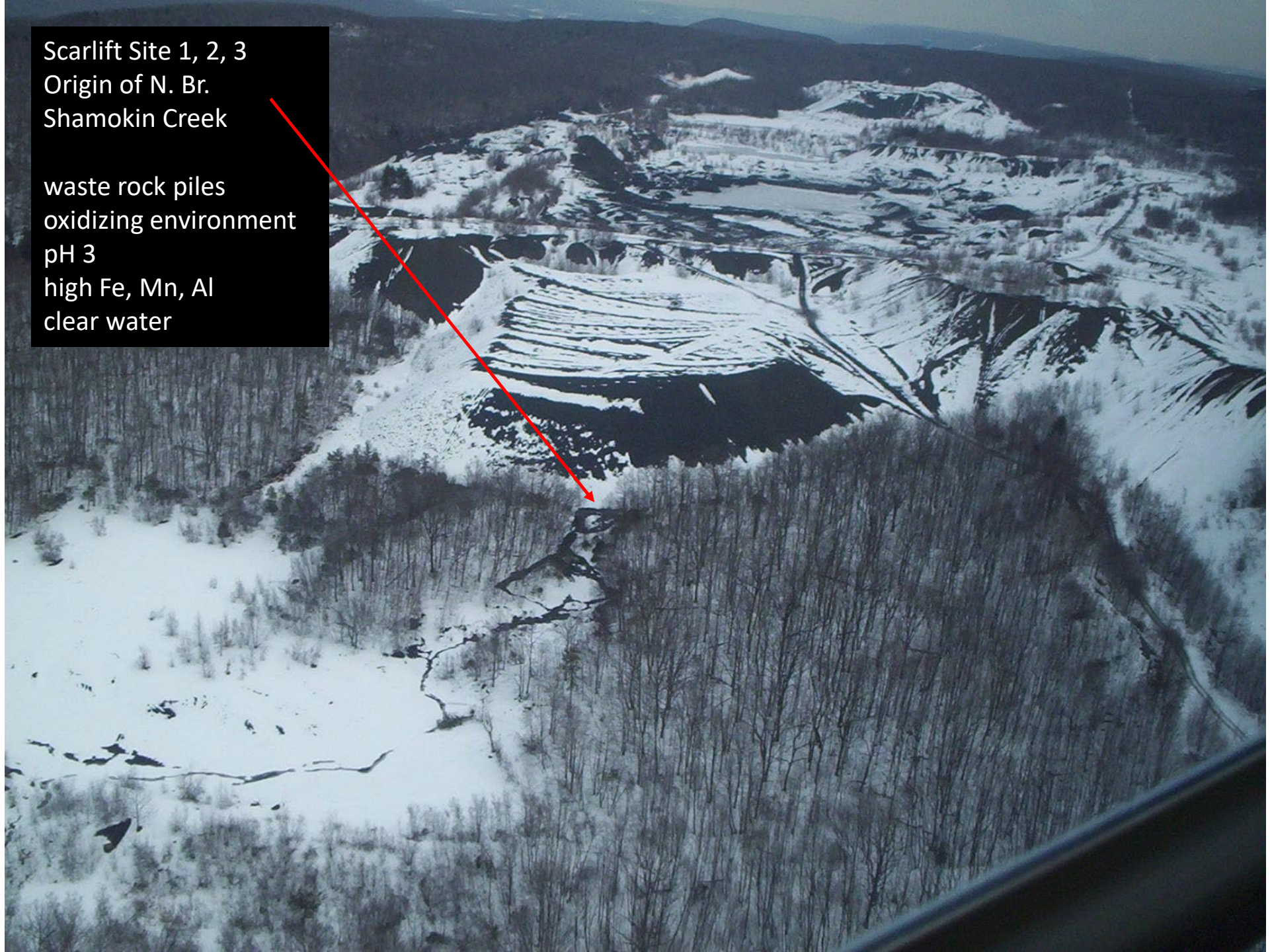
SCRA Site Identification Number <sup>a</sup>	AMD Metals Rank <sup>b</sup>	Net Alkalinity Rank	Sulfate Rank	Percentage Dissolved Fe, Al, and Mn load	Cumulative Percentage Fe, Al, and Mn load	Principal Characteristics <sup>c</sup>	Remedial Alternatives <sup>d</sup>					Comments	Wetland Area, acres <sup>e</sup>		
							Remove Culm bank	VFCW	ALD	OLD	OLC			Acrobic Pond(s)	Active Treatment
19	1	4	1	23.5	23.5	Very large flow; high Fe, Mn; moderate Al; net acidic; suboxic.						1	Passive methods difficult because of physical layout. Consider active treatment using heterogeneous catalysis of iron oxidation by ferric hydroxide. Additional alkalinity needed to balance acidity and facilitate iron oxidation. Potentially could be treated passively south of Route 61.	17.8	
12	2	2	3	17.1	40.6	Very large flow; high Fe, Mn; moderate Al; net acidic; oxic.						?	?	Consider <i>in-situ</i> alkaline addition to present pond (requires introduction of dissolved oxygen into strip pit) or active treatment. PaDEP BAMR considering filling pit to eliminate physical hazard at County's request. (Additional water chemistry and survey data available from C. Kirby in Dept. of Geology, Bucknell University.)	13.1
49	3	43*	2	12.6	53.2	Large flow; high Fe, Mn; moderate Al; net acidic?; anoxic.			1			1	2	May be able to introduce limestone into present pump slope or mine pool through borehole; water must be routed across Carbon Run to ponds. Consider active treatment with no alkaline addition, but using heterogeneous catalysis of iron	10.0

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53	4	1	4	12.3	65.5	Very large flow; very high Fe, Mn; very high Al; net	?					?	1	Discharges from culvert under Route 61 immediately into Shamokin Creek; no area for treatment unless discharge relocated. Large flow, high concentrations of	8.8

Aerial Pictures – headwaters toward mouth

Scarlift Site 1, 2, 3  
Origin of N. Br.  
Shamokin Creek

waste rock piles  
oxidizing environment  
pH 3  
high Fe, Mn, Al  
clear water





Scarlift Site 1, 2, 3  
Origin of N. Br.  
Shamokin Creek

waste rock piles  
oxidizing environment  
pH 3  
High Fe, Mn, Al  
clear water



## Headwaters – surface mine discharges



Brad Jordan at Scarlift Site 2 discharge  
From a large aerated spoils pile

- pH 3
- high Fe, Al, Mn
- net acidic
- clear water

## Headwaters – surface mine & deep mine discharges



Scarlift Site 4  
Luxuriant algae growth – not much living to eat it  
From deep mine & spoils pile

- pH 3.5
- high Fe, Al, Mn
- net acidic
- clear water

## Scarlift Site 4

### Origin of N. Br. Shamokin Creek

- waste rock piles
- mixed oxidizing/reducing environment
- pH 3.7
- High Fe, Mn, Al

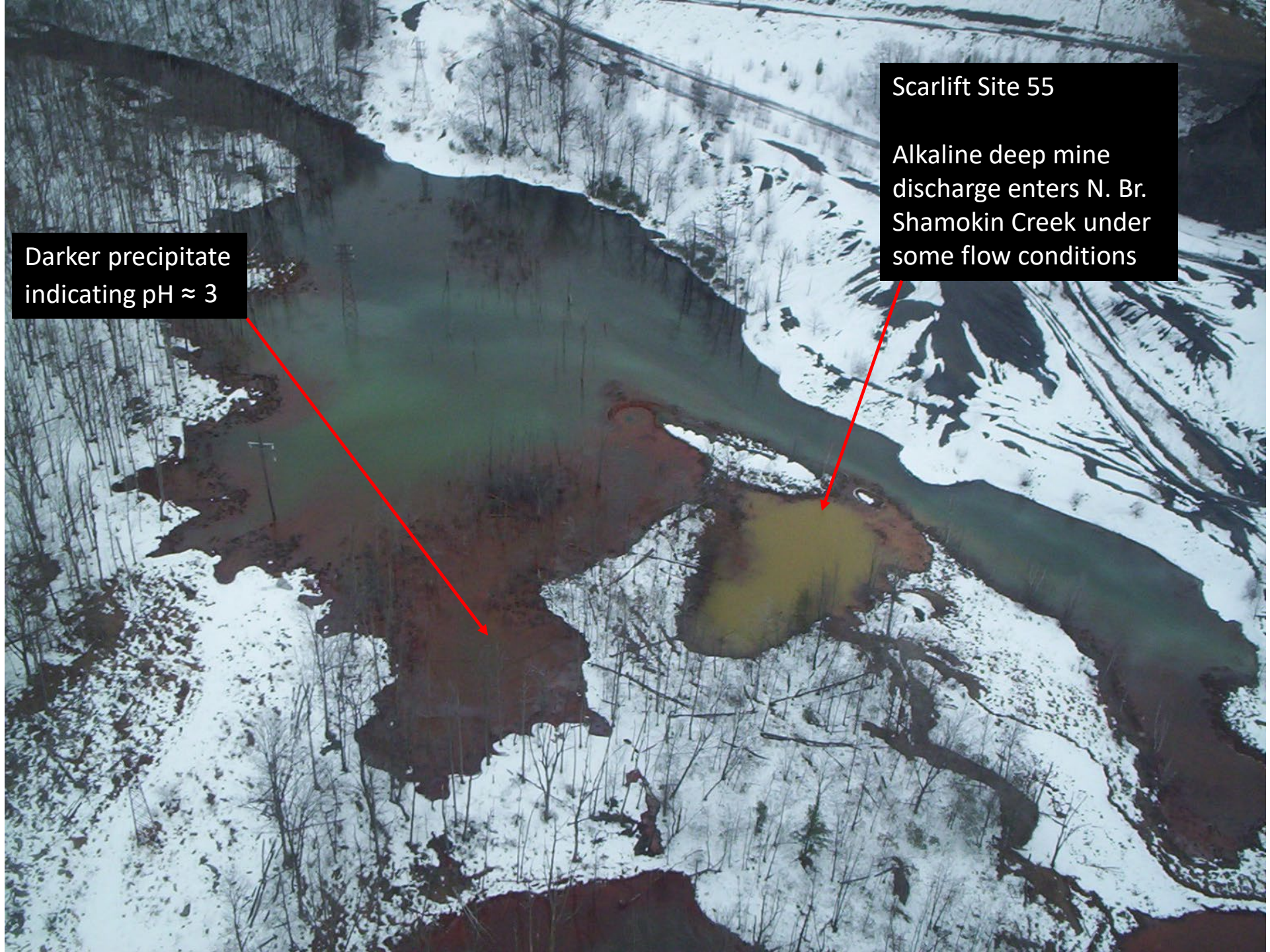


Scarlift Site 5  
N. Br. Shamokin Creek

waste rock piles  
Intermediate redox  
pH 4  
High Fe, Mn, Al




Shamokin Creek



Darker precipitate  
indicating pH  $\approx$  3

Scarlift Site 55  
Alkaline deep mine  
discharge enters N. Br.  
Shamokin Creek under  
some flow conditions



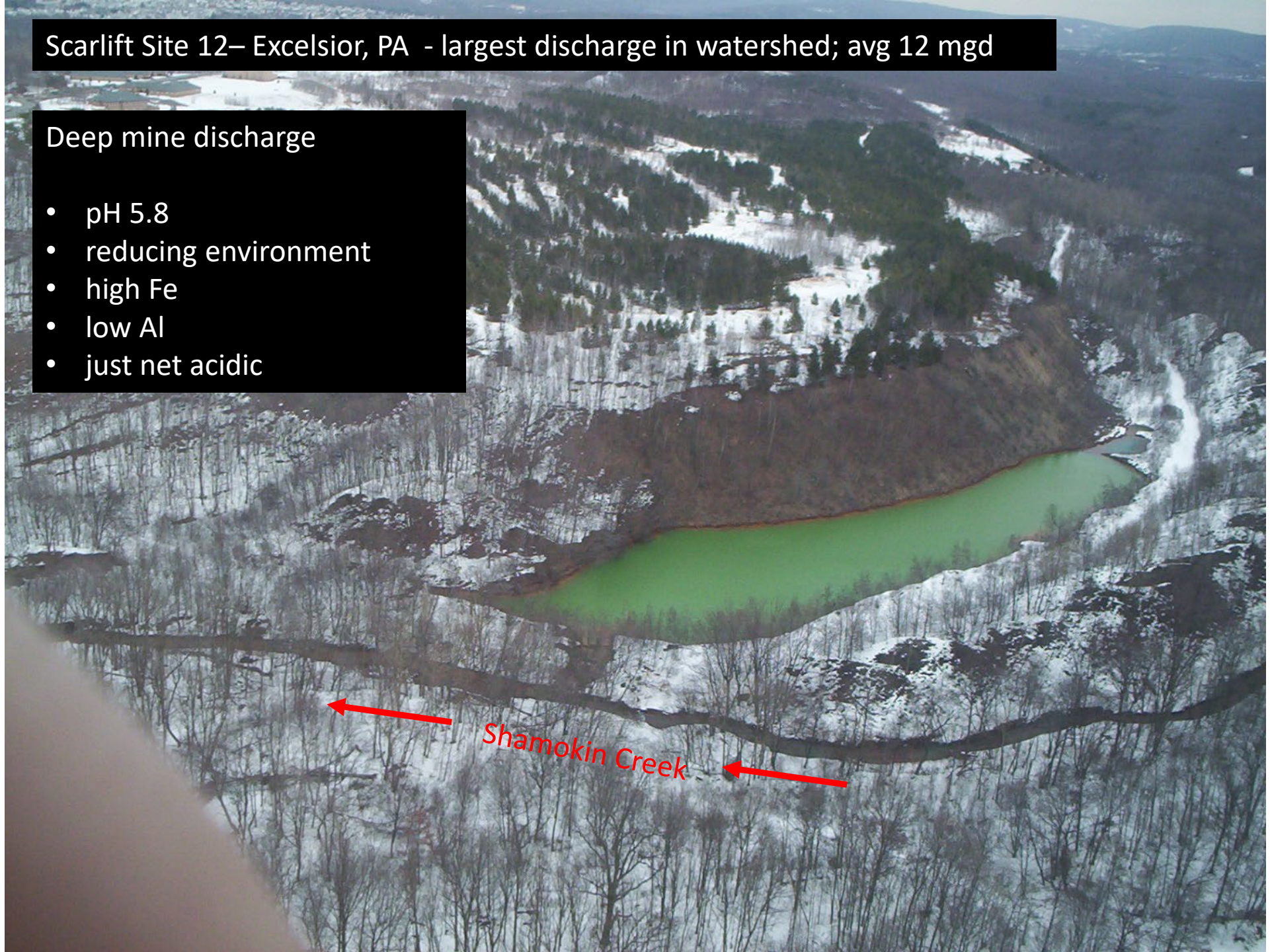
Waste rock piles produce

- oxidizing conditions
- low pH
- high metals

Scarlift Site 12— Excelsior, PA - largest discharge in watershed; avg 12 mgd

### Deep mine discharge

- pH 5.8
- reducing environment
- high Fe
- low Al
- just net acidic



An aerial photograph of a winter forest. The ground is covered in snow, and the trees are mostly bare. A stream flows through the center of the image. On the left side, there is a small, light-colored building. The overall scene is desolate and cold.

## Scarlift Site 23

- pH 3.5
- high Fe, Al
- net acidic



# Scarlift Site 21 – Ranshaw, PA - deep mine discharge

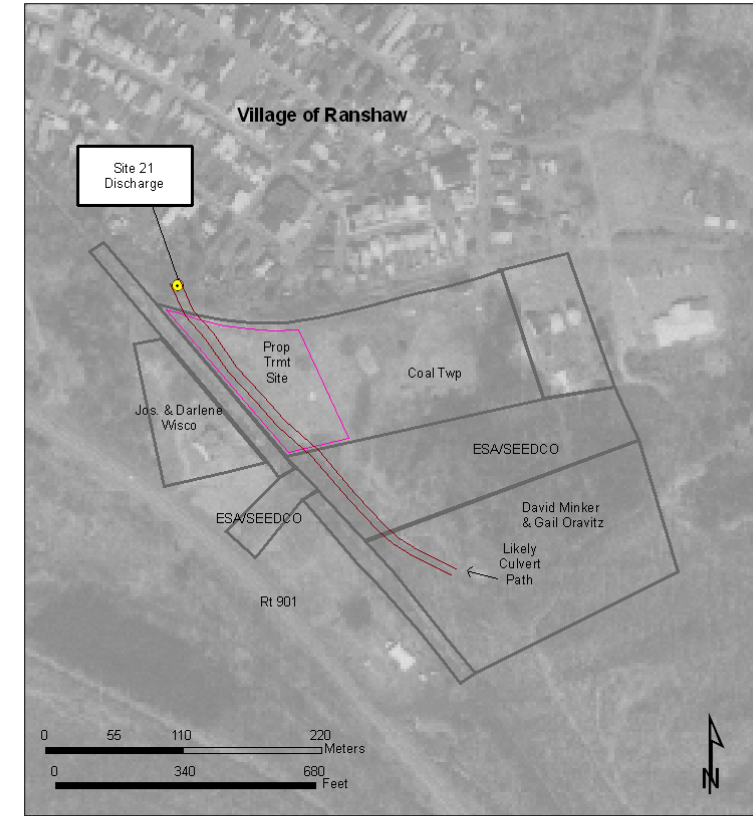


Figure 1. Aerial photograph with land ownership near proposed Site 21 treatment.

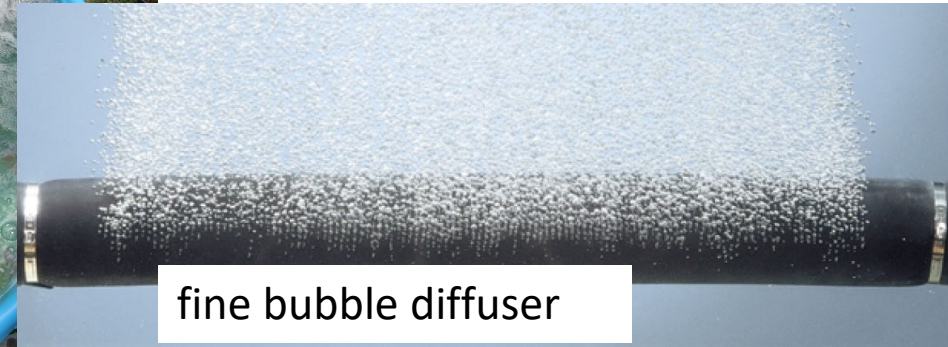


- 0.5 cfs
- pH 6.1
- high Fe
- low Al
- net alkaline

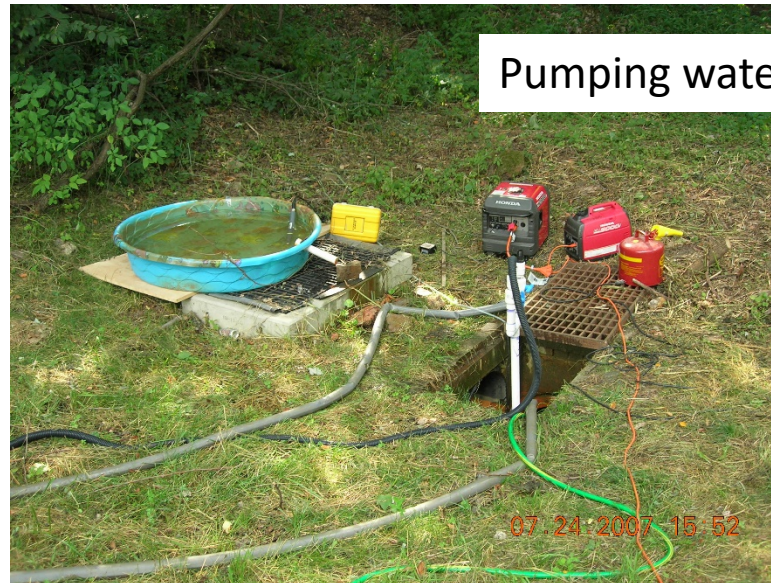
# Scarlift Site 21 – Ranshaw, PA – Fe(II) oxidation rate experiments – stage 2



Could be treated with aeration alone; would require pumping and area to capture solids; no limestone needed



# Scarlift Site 21 – Ranshaw, PA – Fe(II) oxidation rate experiments – stage 3



Scarlift Site 19



10 cfs  
pH 5.9  
high Fe  
low Al

Scarlift Site 20 6/27/2006 flooding



10 cfs  
pH 5.9  
high Fe  
low Al

SCRA-installed weirs

Scarlift Site 49 (pump slope) - Sept 9, 2011



Scarlift Site 36 – 0 (usually) to 2 mgd flow  
Double V-notch weir for flowrate measurement



Scarlift Site 23 – Ranshaw, PA

SCRA-installed weir washed out



Scarlift Site 49 – pump slope



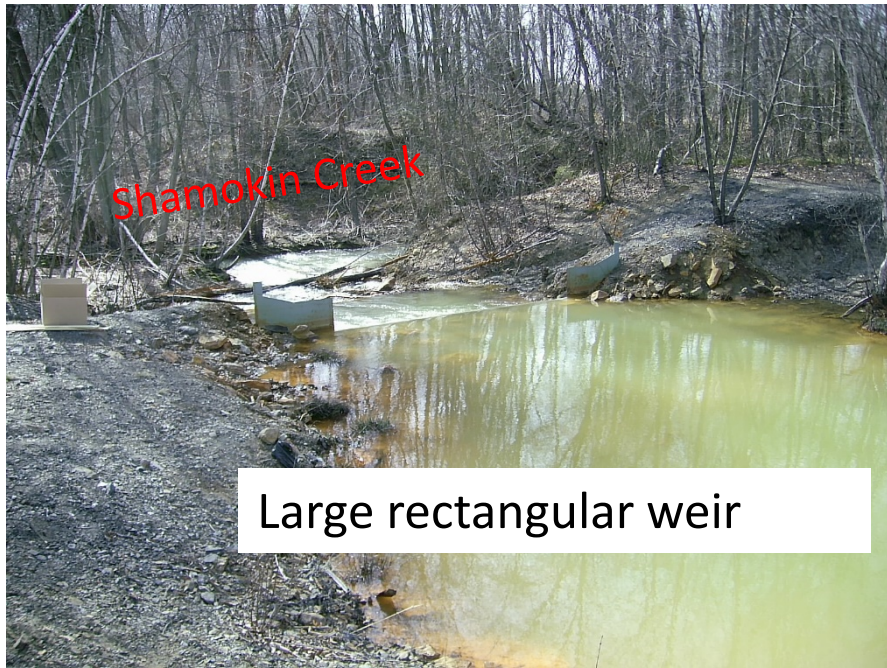
6/27/2006 flooding

Scarlift Site 12– Excelsior, PA - largest discharge in watershed; avg 12 mgd

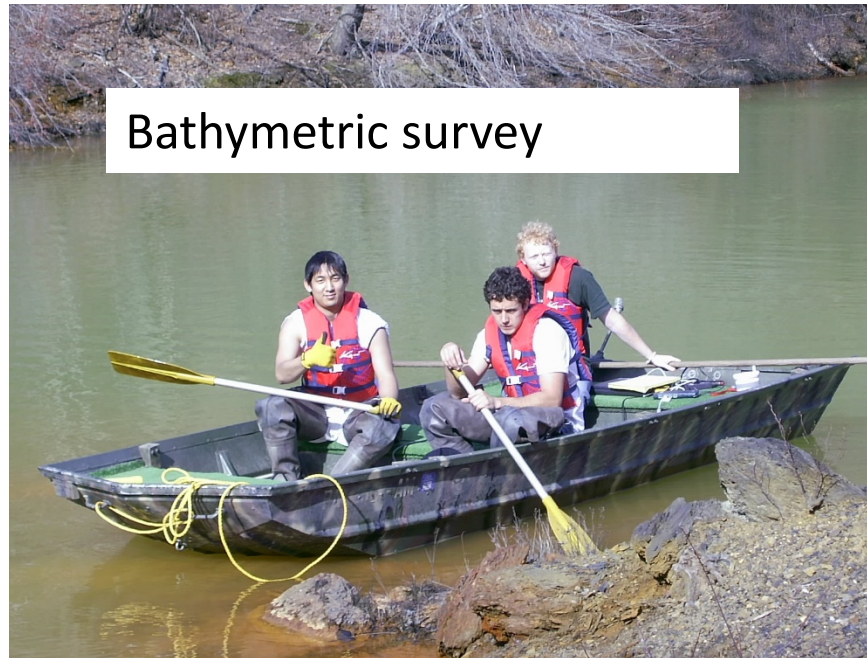


This active mine at least 12 m deep; flooded overnight

- 7 cfs
- pH 5.8
- reducing conditions
- high Fe
- low Al
- just net acidic

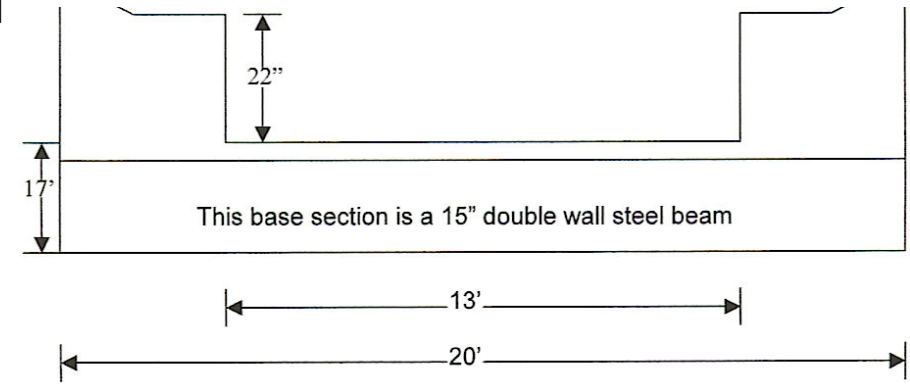


Large rectangular weir



Bathymetric survey

Scarlift Site 12– Excelsior, PA - largest discharge in watershed; avg 12 mgd



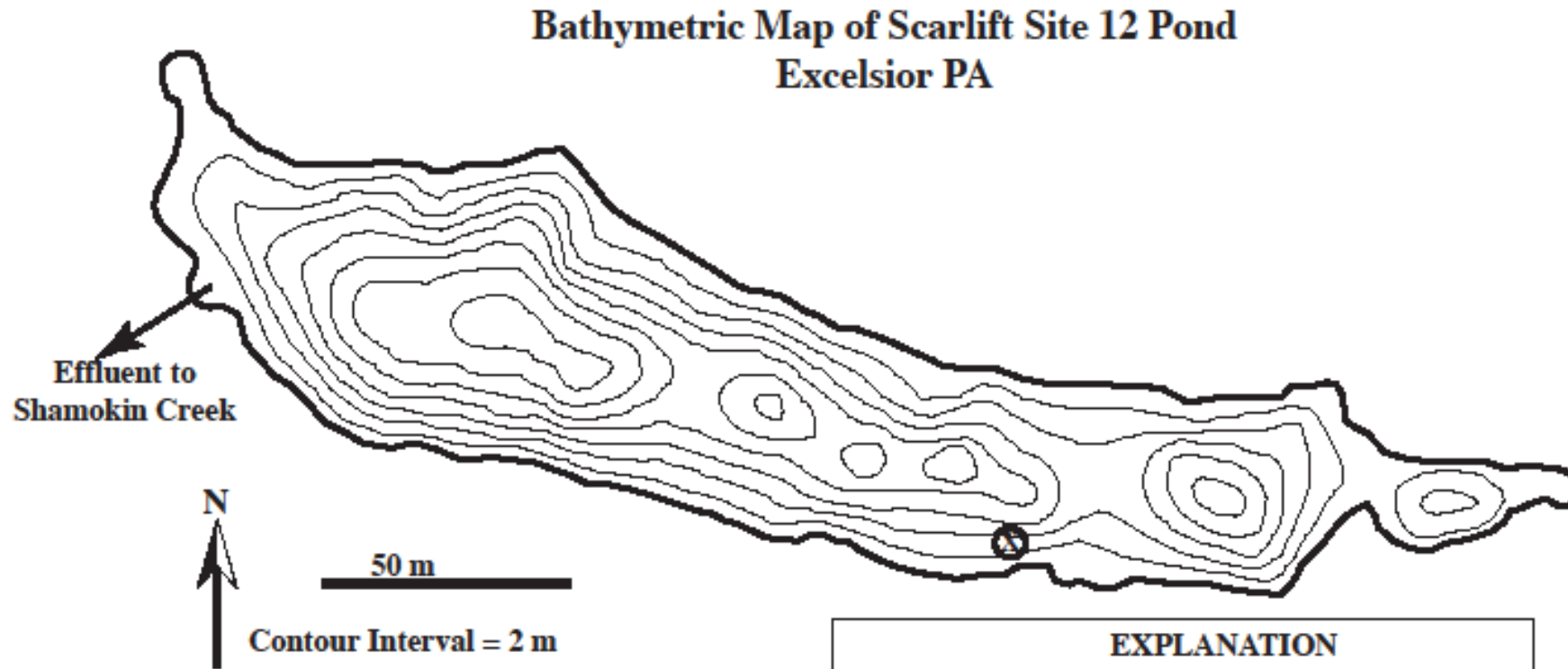
The lip of the weir is a 2" x 1/4" angle iron that was ground to a sharper point.

Site 12 (Excelsior) discharge weir








Scarlift Site 12– Excelsior, PA - largest discharge in watershed; avg 12 mgd



Map by Bucknell Department of Geology:  
Joshua A. Smith, K. Byron Light, Carl S. Kirby, 7/2/01

**EXPLANATION**

-  Pond perimeter (296 m above sea level)
-  Contour lines (kriging; linear variogram)
-  Approx. location of breach in high wall

Volume of pond estimated to be  
 $1.6 \times 10^5 \text{ m}^3$  by ArcView 3.1 TIN

Scarlift Site 51 just upstream of Shamokin Creek  
near spoils pile north of Shamokin



- 1 cfs
- pH 5
- high Fe, Al, Mn
- net acidic



Scarlift Site 23 – Double v-notch weir replacing damaged upstream wooden weir



Metal V-notch weir damaged 6/27/06 flood



Scarlift Site 49 (pump slope leading to deep mine)  
just upstream of Carbon Run



Harriet the  
Geochemistry  
Field Dog  
R.I.P

Scarlift Site 49 (pump slope) - building a rectangular weir for flowrate



Ashland Borehole discharge – south of Shamokin Creek watershed – quite artesian



## Headwaters area

Remining (removal of spoils to burn producing electricity) removes pyritic material, helps prevent AMD





Shamokin Creek 10 miles downstream of the last mine discharge. Color changes often from clear to red depending on rainfall in particular parts of the watershed. The pH ranges from 4 to 6.

Site 42 treatment system Vertical flow wetland  
Now that's dedication to research!







USGS electrofishing Shamokin Creek 2002  
downstream of the last mine discharge

6 species, several individuals



Catfish and suckers



## Shamokin Creek Restoration Alliance

Typical of watershed groups, about 200 members, but about 2 to 5 people do most of the volunteer work

